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# SODIUM BICARBONATE Safety Data Sheet

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#### 1. IDENTIFICATION

**Product name:** Sodium bicarbonate

Synonyms: Sodium hydrogen carbonate; Baking soda; Bicarbonate of soda; Sodium acid carbonate; Carbonic

acid, monosodium salt.

Manufacturer:

Natrium Products, Inc. 58 Pendleton Street Cortland, NY 13045

USA

Telephone numbers:

General inquiries: (607) 753-9829 Emergencies (US and Canada):

CHEMTREC (Customer Number 724993) (800) 424-9300 or 703-527-3887 (collect)

#### Recommended uses:

Food additive; pharmaceutical ingredient; water treatment; raw material for paper and chemical manufacturing; animal feed additive; pH control.

#### 2. HAZARD IDENTIFICATION

There are no appreciable health or environmental effects associated with this material.

Hazard classification: Not classified Label elements: No applicable labeling

Other potential health effects:

**Eyes:** Direct contact may cause irritation due to abrasion.

Skin: Not a skin irritant.

Inhalation: No known effects.

# 3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical name: Sodium hydrogen carbonate Chemical formula: NaHCO<sub>3</sub>

Synonyms: Sodium bicarbonate; Baking soda; Bicarbonate of soda; Sodium acid carbonate; Carbonic acid,

monosodium salt.

**CAS Number:** 144-55-8

Concentration (% by Weight): 100%

#### 4. FIRST AID MEASURES

**Eye contact:** Irrigate with flowing water immediately and continuously for 15 minutes. Consult a physician if

necessary.

**Skin contact:** Wash off in flowing water or shower. If necessary, consult physician.

**Ingestion:** Do not induce vomiting. Seek medical attention immediately if overdose is taken.

Note to physician: Large doses, particularly in patients with renal insufficiency, have produced systemic alkalosis and/or expansion in the extra-cellular fluid volume with edema.

Inhalation: Remove to fresh air. Seek medical attention if discomfort persists.

## 5. FIRE-FIGHTING MEASURES

Product is non-combustible. Thermal decomposition products are carbon dioxide and sodium carbonate (soda ash). Carbon dioxide is an asphyxiant, and soda ash is an irritant.

**Protective equipment:** Self- contained breathing apparatus is necessary if large quantities are involved.

Extinguishing media: Use extinguishing material that is appropriate for fire in the surrounding area.



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# **6. ACCIDENTAL RELEASE MEASURES**

Sweep up into clean, dry containers for salvage or disposal. Wash away uncontaminated residue with water.

#### 7. HANDLING AND STORAGE

Avoid contact with eyes and skin. Keep separated from acids. Store in a cool, dry place.

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

**Exposure limits:** Not established.

**Engineering controls:** Provide general and/or local exhaust ventilation to control airborne dust.

**Personal Protection:** 

Eyes & Face: Safety glasses for dusty conditions.

**Respiratory:** NIOSH approved dust mask.

Miscellaneous: Full cover clothing, general purpose gloves.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

**Appearance:** White crystalline powder or granules.

Flammability: None.

Upper/lower flammability/explosive limits: Not applicable.

Odor: None.

Odor threshold: Not applicable.

Vapor pressure: Not applicable.

Vapor density: Not applicable.

pH of 0.1 M solution (0.84% w/v): 8.3 @ 25°C

**Density:** 2.2 g/cm<sup>3</sup>.

Melting point: Not applicable (thermal decomposition occurs on heating).

Solubility in water: 86 g/L @ 20°C.

**Boiling point:** Not applicable. **Flash point:** Not applicable.

Evaporation rate: Not applicable.

Partition coefficient, n-octanol/water: No data available.

Auto-ignition temperature: Not applicable.

**Decomposition temperature:** Starts to decompose when heated above 50°C (122°F).

Viscosity: Not applicable.

# 10. STABILITY AND REACTIVITY

Reactivity: Hazardous reactions or polymerization will not occur under normal conditions.

Chemical stability: Stable under recommended handling and storage conditions. (See Section 7.)

Conditions to avoid: Temperatures above 50°C (122°F).

Incompatible materials: Reacts with acids, releasing carbon dioxide.

Hazardous decomposition products: Carbon dioxide and sodium carbonate (soda ash).

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# 11. TOXICOLOGICAL INFORMATION

Acute Oral:  $LD_{50}$  (rat) > 4000 mg/kg. Acute Inhalation:  $LC_{50}$  (rat) > 4.74 mg/L.

Eyes: Minimally irritating (rabbit, EPA TSCA 40 CFR 798.4500); Irritating (rabbit, Draize test, dose of 220 mg).

Skin: Slightly irritating (rabbit).

Carcinogenicity: Not listed as a carcinogen or potential carcinogen by the National Toxicology Program (NTP),

the International Agency for Research on Cancer (IARC), or the U.S. Occupational Safety and Health

Administration (OSHA).

# 12. ECOLOGICAL INFORMATION

#### Aquatic toxicity:

Fish: LC50 = 7700 mg/L (Rainbow trout, 96-hr. exposure). Fish: LC50 = 7100 mg/L (Bluegill sunfish, 96-hr. exposure).

Invertebrates: EC50 > 1000 mg/L (Daphnia magna, 48-hr. exposure).

Persistence/Bioaccumulation potential: Not expected to persist or bioaccumulate in the environment.

**Biodegradation:** Not applicable.

Mobility: High potential for movement from soil to groundwater is expected based on aqueous solubility.

#### 13. DISPOSAL CONSIDERATIONS

Not a hazardous material. Dispose in a landfill in accordance with pertinent federal, state and local regulations. Empty containers may be incinerated or discarded as ordinary waste.

#### 14. TRANSPORT INFORMATION

Not regulated by the U.S. Department of Transportation.

#### 15. REGULATORY INFORMATION

CERCLA (40 CFR 302.4): Not a hazardous substance.

RCRA (40 CFR 261): Not a hazardous waste.

TSCA (40 CFR 710): Listed.

OSHA (29 CFR 1910.1200): Not hazardous.

SARA. Title III Sections 302 (40 CFR 355), 313 (40 CFR 372): Not a hazardous or toxic chemical.

European Inventory (EINECS): 205-633-8.

Japanese Inventory (MITI): 1-164.

U.S. Food and Drug Administration: Generally recognized as safe (GRAS) direct food additive

(21 CFR 184.1736).

# **16. OTHER INFORMATION**

Maximum use level for drinking water corrosion and scale control: 100mg/L per NSF/ANSI 60 - 2014a.

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